

CHEMISTRY GELIJKINGEN

1. $\dots \text{H}_2 + \dots \text{O}_2 \rightarrow \dots \text{H}_2\text{O}$	2. $\dots \text{Na} + \dots \text{Cl}_2 \rightarrow \dots \text{NaCl}$
3. $\dots \text{H}_2 + \dots \text{N}_2 \rightarrow \dots \text{NH}_3$	4. $\dots \text{Al} + \dots \text{O}_2 \rightarrow \dots \text{Al}_2\text{O}_3$
5. $\dots \text{K} + \dots \text{Br}_2 \rightarrow \dots \text{KBr}$	6. $\dots \text{KClO}_3 \rightarrow \dots \text{KCl} + \dots \text{O}_2$
7. $\dots \text{P} + \dots \text{Cl}_2 \rightarrow \dots \text{PCl}_3$	8. $\dots \text{C}_3\text{H}_8 + \dots \text{O}_2 \rightarrow \dots \text{H}_2\text{O} + \dots \text{CO}_2$
9. $\dots \text{P}_2\text{O}_5 \rightarrow \dots \text{P} + \dots \text{O}_2$	10. $\dots \text{CS}_2 + \dots \text{O}_2 \rightarrow \dots \text{CO}_2 + \dots \text{SO}_2$
11. $\dots \text{Sb}_2\text{O}_5 + \dots \text{HCl} \rightarrow \dots \text{SbCl}_5 + \dots \text{H}_2\text{O}$	12. $\dots \text{Al} + \dots \text{O}_2 \rightarrow \dots \text{Al}_2\text{O}_3$
13. $\dots \text{FeCl}_3 + \dots \text{Na} \rightarrow \dots \text{Fe} + \dots \text{NaCl}$	14. $\dots \text{C}_5\text{H}_{10} + \dots \text{O}_2 \rightarrow \dots \text{CO}_2 + \dots \text{H}_2\text{O}$
15. $\dots \text{Li} + \dots \text{O}_2 \rightarrow \dots \text{Li}_2\text{O}$	16. $\dots \text{Ga} + \dots \text{F}_2 \rightarrow \dots \text{GaF}_3$
17. $\dots \text{As}_2\text{O}_5 + \dots \text{HBr} \rightarrow \dots \text{AsBr}_5 + \dots \text{H}_2\text{O}$	18. $\dots \text{CBr}_2 + \dots \text{O}_2 \rightarrow \dots \text{CO}_2 + \dots \text{BrO}_2$

19. $\dots \text{S}_4\text{H}_8 + \dots \text{O}_2 \rightarrow \dots \text{SO}_2 + \dots \text{H}_2\text{O}$
21. $\dots \text{N}_3\text{H}_8 + \dots \text{Br}_2 \rightarrow \dots \text{NBr}_2 + \dots \text{H}_2\text{B}$
23. $\dots \text{NF}_3 + \dots \text{NO} \rightarrow \dots \text{N}_2 + \dots \text{F}_2\text{O}$
25. $\dots \text{N}_5\text{H}_{10} + \dots \text{O}_2 \rightarrow \dots \text{NO}_2 + \dots \text{H}_2\text{O}$
27. $\dots \text{C}_3\text{Na}_6 + \dots \text{F}_2 \rightarrow \dots \text{CF}_2 + \dots \text{Na}_2\text{F}$
29. $\dots \text{K}_2\text{O} + \dots \text{H}_2\text{O} \rightarrow \dots \text{KOH}$
31. $\dots \text{SiC} + \dots \text{Cl}_2 \rightarrow \dots \text{SiCl}_4 + \dots \text{C}$
33. $\dots \text{Na} + \dots \text{H}_2\text{O} \rightarrow \dots \text{NaOH} + \dots \text{H}_2$
35. $\dots \text{MgNH}_4\text{PO}_4 \rightarrow \dots \text{Mg}_2\text{P}_2\text{O}_7 + \dots \text{NH}_3 + \dots \text{H}_2\text{O}$

