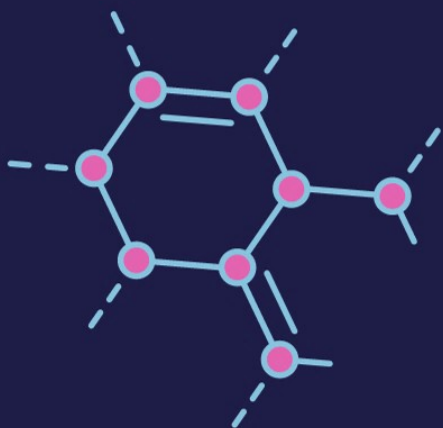


KAARTJES BEAT THE CHEMISTRY REACTIEVERGELIJKINGEN



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| 1. $\dots \text{H}_2 + \dots \text{O}_2 \rightarrow \dots \text{H}_2\text{O}$ | 2. $\dots \text{Na} + \dots \text{Cl}_2 \rightarrow \dots \text{NaCl}$ |
| 3. $\dots \text{H}_2 + \dots \text{N}_2 \rightarrow \dots \text{NH}_3$ | 4. $\dots \text{Al} + \dots \text{O}_2 \rightarrow \dots \text{Al}_2\text{O}_3$ |
| 5. $\dots \text{K} + \dots \text{Br}_2 \rightarrow \dots \text{KBr}$ | 6. $\dots \text{KClO}_3 \rightarrow \dots \text{KCl} + \dots \text{O}_2$ |
| 7. $\dots \text{P} + \dots \text{Cl}_2 \rightarrow \dots \text{PCl}_3$ | 8. $\dots \text{C}_3\text{H}_8 + \dots \text{O}_2 \rightarrow \dots \text{H}_2\text{O} + \dots \text{CO}_2$ |
| 9. $\dots \text{P}_2\text{O}_3 \rightarrow \dots \text{P} + \dots \text{O}_2$ | 10. $\dots \text{CS}_2 + \dots \text{O}_2 \rightarrow \dots \text{CO}_2 + \dots \text{SO}_2$ |
| 11. $\dots \text{Sb}_2\text{O}_5 + \dots \text{HCl} \rightarrow \dots \text{SbCl}_5 + \dots \text{H}_2\text{O}$ | 12. $\dots \text{Al} + \dots \text{O}_2 \rightarrow \dots \text{Al}_2\text{O}_3$ |
| 13. $\dots \text{FeCl}_3 + \dots \text{Na} \rightarrow \dots \text{Fe} + \dots \text{NaCl}$ | 14. $\dots \text{C}_5\text{H}_{10} + \dots \text{O}_2 \rightarrow \dots \text{CO}_2 + \dots \text{H}_2\text{O}$ |
| 15. $\dots \text{Li} + \dots \text{O}_2 \rightarrow \dots \text{Li}_2\text{O}$ | 16. $\dots \text{Ga} + \dots \text{F}_2 \rightarrow \dots \text{GaF}_3$ |
| 17. $\dots \text{As}_2\text{O}_5 + \dots \text{HBr} \rightarrow \dots \text{AsBr}_5 + \dots \text{H}_2\text{O}$ | 18. $\dots \text{CBr}_2 + \dots \text{O}_2 \rightarrow \dots \text{CO}_2 + \dots \text{Br}_2\text{O}$ |

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| 19. $\dots \text{S}_4\text{H}_8 + \dots \text{O}_2 \rightarrow \dots \text{SO}_2 + \dots \text{H}_2\text{O}$ | 20. $\dots \text{S}_4\text{H}_8 + \dots \text{O}_2 \rightarrow \dots \text{SO}_2 + \dots \text{H}_2\text{O}$ |
| 21. $\dots \text{N}_3\text{H}_8 + \dots \text{Br}_2 \rightarrow \dots \text{NBr}_2 + \dots \text{H}_2\text{Br}$ | 22. $\dots \text{Al} + \dots \text{HBr} \rightarrow \dots \text{AlBr}_3 + \dots \text{H}_2$ |
| 23. $\dots \text{NF}_3 + \dots \text{NO} \rightarrow \dots \text{N}_2 + \dots \text{F}_2\text{O}$ | 24. $\dots \text{S}_3\text{H}_6 + \dots \text{O}_2 \rightarrow \dots \text{SO}_2 + \dots \text{H}_2\text{O}$ |
| 25. $\dots \text{N}_5\text{H}_{10} + \dots \text{O}_2 \rightarrow \dots \text{NO}_2 + \dots \text{H}_2\text{O}$ | 26. $\dots \text{H}_2\text{Se} + \dots \text{O}_2 \rightarrow \dots \text{SeO}_2 + \dots \text{H}_2\text{O}$ |
| 27. $\dots \text{C}_3\text{N}_6 + \dots \text{F}_2 \rightarrow \dots \text{CF}_2 + \dots \text{Na}_2\text{F}$ | 28. $\dots \text{C}_6\text{H}_6 + \dots \text{O}_2 \rightarrow \dots \text{CO}_2 + \dots \text{H}_2\text{O}$ |
| 29. $\dots \text{K}_2\text{O} + \dots \text{H}_2\text{O} \rightarrow \dots \text{KOH}$ | 30. $\dots \text{Sb} + \dots \text{O}_2 \rightarrow \dots \text{Sb}_4\text{O}_6$ |
| 31. $\dots \text{SiC} + \dots \text{Cl}_2 \rightarrow \dots \text{SiCl}_4 + \dots \text{C}$ | 32. $\dots \text{Mg} + \dots \text{N}_2 \rightarrow \dots \text{Mg}_3\text{N}_2$ |
| 33. $\dots \text{Na} + \dots \text{H}_2\text{O} \rightarrow \dots \text{NaOH} + \dots \text{H}_2$ | 34. $\dots \text{KClO}_3 \rightarrow \dots \text{KCl} + \dots \text{O}_2$ |
| 35. $\dots \text{MgNH}_4\text{PO}_4 \rightarrow \dots \text{Mg}_2\text{P}_2\text{O}_7 + \dots \text{NH}_3 + \dots \text{H}_2\text{O}$ | 36. $\dots \text{NH}_3 + \dots \text{NO} \rightarrow \dots \text{N}_2 + \dots \text{H}_2\text{O}$ |